

# Periodic Table of the Elements

1 Hydrogen H 1.01	2 Helium He 4.00
3 Lithium Li 6.94	4 Beryllium Be 9.01
11 Sodium Na 22.99	12 Magnesium Mg 24.31
19 Potassium K 39.10	20 Calcium Ca 40.08
37 Rubidium Rb 85.47	38 Strontium Sr 87.62
55 Cesium Cs 132.91	56 Barium Ba 137.33
87 Francium Fr [223]	88 Radium Ra [226]

metal

Roman numerals

21 Scandium Sc 44.96	22 Titanium Ti 47.88	23 Vanadium V 50.94	24 Chromium Cr 52.00	25 Manganese Mn 54.94	26 Iron Fe 55.85	27 Cobalt Co 58.93	28 Nickel Ni 58.69	29 Copper Cu 63.55	30 Zinc Zn 65.39	31 Gallium Ga 69.72	32 Germanium Ge 72.61	33 Arsenic As 74.92	34 Selenium Se 78.96	35 Bromine Br 79.90	36 Krypton Kr 83.80
39 Yttrium Y 88.91	40 Zirconium Zr 91.22	41 Niobium Nb 92.91	42 Molybdenum Mo 95.94	43 Technetium Tc [98]	44 Ruthenium Ru 101.07	45 Rhodium Rh 102.91	46 Palladium Pd 106.42	47 Silver Ag 107.87	48 Cadmium Cd 112.41	49 Indium In 114.82	50 Tin Sn 118.71	51 Antimony Sb 121.76	52 Tellurium Te 127.60	53 Iodine I 126.90	54 Xenon Xe 131.29
71 Lutetium Lu 174.97	72 Hafnium Hf 178.49	73 Tantalum Ta 180.95	74 Tungsten W 183.84	75 Rhenium Re 186.21	76 Osmium Os 190.23	77 Iridium Ir 192.22	78 Platinum Pt 195.08	79 Gold Au 196.97	80 Mercury Hg 200.59	81 Thallium Tl 204.38	82 Lead Pb 207.20	83 Bismuth Bi 208.98	84 Polonium Po [209]	85 Astatine At [210]	86 Radon Rn [222]
103 Lawrencium Lr [262]	104 Rutherfordium Rf [261]	105 Dubnium Db [262]	106 Seaborgium Sg [266]	107 Bohrium Bh [264]	108 Hassium Hs [269]	109 Meitnerium Mt [268]	110 Darmstadtium Ds [271]	111 Roentgenium Rg [272]	112 Copernicium Cn [277]	113 Nihonium Nh [284]	114 Flerovium Fl [289]	115 Moscovium Mc [288]	116 Livermorium Lv [293]	117 Tennessine Ts [294]	118 Oganesson Og [294]

3+ 4+ 3- 2- 1-  
nonmetals

57 Lanthanum La 138.91	58 Cerium Ce 140.12	59 Praseodymium Pr 140.91	60 Neodymium Nd 144.24	61 Promethium Pm [145]	62 Samarium Sm 150.36	63 Europium Eu 151.97	64 Gadolinium Gd 157.25	65 Terbium Tb 158.93	66 Dysprosium Dy 162.50	67 Holmium Ho 164.93	68 Erbium Er 167.26	69 Thulium Tm 168.93	70 Ytterbium Yb 173.04
89 Actinium Ac [227]	90 Thorium Th 232.04	91 Protactinium Pa 231.04	92 Uranium U 238.03	93 Neptunium Np [237]	94 Plutonium Pu [244]	95 Americium Am [243]	96 Curium Cm [247]	97 Berkelium Bk [247]	98 Californium Cf [251]	99 Einsteinium Es [252]	100 Fermium Fm [257]	101 Mendelevium Md [258]	102 Nobelium No [259]



# Covalent Compounds

- 2 nonmetals
- use prefixes
- no charges to crisscross

## Prefixes

1 mono

2 di

3 tri

4 tetra

5 penta

6 hexa

7 hepta

8 octa

9 nona

10 deca

ex) sulfur trioxide



dinitrogen  
pentoxide



NBr<sub>3</sub>  
nitrogen  
tribromide

P<sub>4</sub>O<sub>10</sub>  
tetraphosphorous  
decoxide



# Ionic Compounds

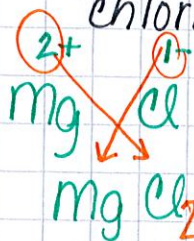
Metal & Nonmetal

or  
Metal & Polyatomic Ion

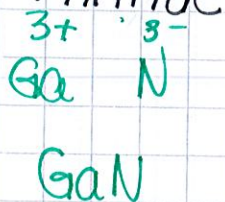
## Formulas

- write symbol & charge of the metal
- do the same with nonmetal/polyatomic ion
- crisscross charges
- simplify

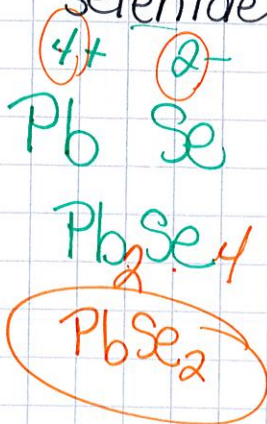
Ex) magnesium chloride



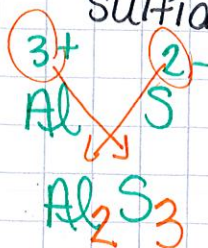
gallium nitride



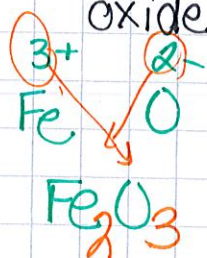
lead (IV) selenide



aluminum sulfide



iron (III) oxide

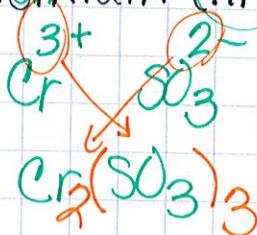


plumbum

calcium nitrate



chromium (III) sulfite



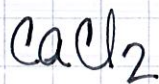


## Names

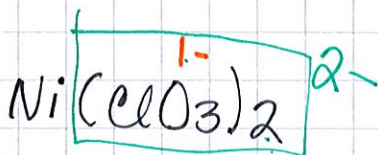
- write name of metal
- if it's a transition metal, find the original charge - write a roman numeral
- write name of nonmetal, change end to -ide or
- write name of the polyatomic ion



lithium  
sulfide



calcium  
chloride



nickel (II)  
chlorate

